# Antitumour polycyclic acridines. Part 10. ${ }^{1}$ Synthesis of penta- and hexa-cyclic heteroaromatic systems by radical cyclisations of substituted 9 -anilinoacridines 

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9-Anilinoacridines substituted with a bromine atom in the 2-position of the anilino group or the 1-position of the acridine moiety can be cyclised with tributyltin hydride-AIBN to penta- or hexacyclic acridines. Of the polycyclic systems 13,14 -dihydropyrrolo $\left[3^{\prime}, 2^{\prime}, 1^{\prime}: 8,1\right]$ quino $[4,3,2-k /]$ acridine 14 a is the most potent cytotoxic agent displaying a mean $\mathrm{GI}_{50}$ concentration against a panel of 60 human tumour cell lines of $0.06 \mu \mathrm{M}$.

## Introduction

In earlier parts of this series we have explored three routes to polycyclic acridines which are initiated by the interactions of 9 -azidoacridine with alkynes, ${ }^{2}$ methylenic compounds ${ }^{3}$ and phosphorus ylides. ${ }^{1,4}$ These syntheses, and others, ${ }^{5}$ generate substituted 9 -( 1 H -1,2,3-triazol-1-yl)acridines which undergo thermal extrusion of nitrogen (Graebe-Ullmann reaction) ${ }^{6}$ to afford pyrido- and quino-acridine systems. ${ }^{1,5}$

The mechanism of the thermolysis ${ }^{2,3}$ and photolysis ${ }^{7}$ of triazolylacridines may involve radical, carbene or dipolar reactive species. However, the reported cyclisation of 9-(2-iodoanilino)acridine $\mathbf{1 a}$ to the quino[4,3,2-kl] acridine $\mathbf{2}$ is undoubtedly a radical process. ${ }^{5}$ We now report further examples of the radical approach to the synthesis of polycyclic acridines.

## Results and discussion

9-(2-Bromoanilino)acridine 1b cyclised to pentacycle 2 with tributyltin hydride-AIBN in boiling toluene. A maximum yield of $31 \%$ of $\mathbf{2}$ was isolated, together with starting material. The inefficiency of cyclisation is probably due to the amino group of the aniline since when $\mathbf{1 b}$ was converted to its anion with sodium hydride-THF and then alkylated with alkyl iodides, the resultant $N$-alkylacridines 3 a-c cyclised with tributyltin hydride-AIBN to the 8 -alkylquinoacridines $\mathbf{4 a - c}$ in 69,79 and $54 \%$ yields, respectively (Scheme 1). The 8 -methylquinoacridine 4a has also been prepared ( $93 \%$ ) by methylating the unsubstituted quinoacridine $\mathbf{2}$ with sodium hydride-dimethyl sulfate, ${ }^{8}$ or by a radical cyclisation approach from the copper(I) iodide oxidation of the anion of a 9-(benzotriazolyl-substituted)-10 methylacridine. ${ }^{9}$ Protecting the anilino NH as a tertiary methylaniline was also an effective strategy to improve the yields of these cyclisations. Thus, 9 -chloroacridine 6 a was reacted in DMF with the anion of 2-bromo- $N$-methylaniline (generated with sodium hydride) to yield 7a. This was then cyclised with tributyltin hydride-AIBN via the intermediate radicals $\mathbf{8}$ and 9 to furnish the 13-methylquinoacridine $10(50 \%)$, together with starting material ( $12 \%$ ) and the debrominated product 9 -( $N$-methylanilino)acridine ( $8 \%$ ), derived from 8 by hydrogen abstraction. A bromo group at the 1-position on the acridine ring can also be used as a source of radicals. Thus, 1-bromo-9( $N$-methylanilino)acridine 7b, formed ( $80 \%$ ) from 1-bromo-9chloroacridine 6b and $N$-methylaniline in HMPA at $110{ }^{\circ} \mathrm{C}$,
cyclised to the same 13-methylquinoacridine $\mathbf{1 0}$ with tributyltin hydride-AIBN in 56\% yield (via radicals 11 and 12) (Scheme 2).
The aforementioned syntheses of the isomeric methylquinoacridines $\mathbf{4 a}$ and $\mathbf{1 0}$ unambiguously confirm their structures. Although not directly comparable because of the different solvents used, the methyl group of the 8-methylquinoacridine 4 a absorbed at $\delta 3.70$ in $\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}$ and at $\delta 4.09$ (in $\mathrm{CDCl}_{3}$ ) for the 13-methylquinoacridine $\mathbf{1 0}$. These isomers can also be distinguished by their characteristic UV-visible spectra (see later).
The radical cyclisation route can be adopted for the synthesis of hexacyclic acridines (Scheme 3). Model compounds 13a, b were synthesised from 9 -chloroacridine and indoline in methanol at $25^{\circ} \mathrm{C}$, or tetrahydroquinoline in HMPA at $110^{\circ} \mathrm{C}$, respectively. The sodio derivative of 7 -iodoindoline was also reacted with 9 -chloroacridine $\mathbf{6 a}$ in DMF to afford 9-(7-iodoindolin-1-yl)acridine $\mathbf{1 3 c}$ which was then cyclised efficiently to the hexacycle 14a with tributyltin hydride-AIBN ( $62 \%$ ). 1-Bromo-9-chloroacridine $\mathbf{6 b}$ was similarly converted to the acridines 13d, e with indoline and tetrahydroquinoline and these intermediates were then cyclised to hexacycles $\mathbf{1 4 a}$ and $\mathbf{b}$, respectively. In contrast, attempted radical cyclisation of the 6-nitroindolinylacridine $\mathbf{1 3 f}$ was unsuccessful. Apart from recovered starting material, the only identified product was an amine, tentatively identified (MS) as the 6 -aminoindoline 13g. It is known that nitro compounds are not good substrates for radical transformations with tributyltin hydride-AIBN. ${ }^{10}$

The reaction between 1-bromo-9-chloroacridine $\mathbf{6 b}$ and racemic 2-methylindoline $\mathbf{1 5}$ to yield the corresponding indolinylacridine 16 ( $78 \%$ ) and subsequent radical cyclisation to the racemic hexacycle 17 appeared to proceed normally (Scheme 4). However, the ${ }^{1} \mathrm{H}$ NMR spectrum of ' $\mathbf{1 6}$ ' showed the presence of two doublets for methyl groups at $\delta 0.91$ and 1.21 in a ratio $2: 1$. This confirmed the presence of two diastereomers in the mixture with the additional source of chirality coming from restricted rotation (atropisomerism) about the pivotal indolineacridine bond.
The differing chromophores in the polycyclic acridines are reflected in their characteristic UV-visible spectra. The long wavelength absorption of the unsubstituted pentacycle $\mathbf{2}$ at $\lambda_{\text {max }}$ 443 nm in ethanol undergoes a bathochromic shift to 488 nm on addition of $\mathrm{HCl}\left(\Delta \lambda_{\max } 45 \mathrm{~nm}\right)$. The $N$-alkylated compounds $4 \mathbf{a}-\mathbf{c}$ display comparable shifts on acidification ( $\Delta \lambda_{\text {max }} 54-56$ nm ) except that the free bases absorb at lower wavelength


Scheme 1


Scheme 2
$R^{1} \equiv R^{2}=H . n=t$
14a: $\quad n=t$
$R^{1}=R^{2}=H, n=2$
b: $\quad n=2$
$R^{1}=H, R^{2}=T-1, n=t$
$\mathrm{R}^{1}=\mathrm{Br} ; \mathrm{R}^{2}=\mathrm{H}: I=1$
$\mathrm{R}^{1}=\mathrm{Br}: \mathrm{R}^{2}=\mathrm{H}, \mathrm{II}=2$
$R^{1}=\mathrm{Br}, R^{2}=6-\mathrm{NO}_{2}, n=1$
$R^{1}=\mathrm{Br}: \mathrm{R}^{2}=6-\mathrm{NH}_{2}: n=\mathrm{t}$
Scheme 3
$6 b$
15
18
19
striking COMPARE ${ }^{13}$ correlations with other topoisomerase II
inhibitors (data not shown).
Melting points were obtained on a Gallenkamp melting point
apparatus and are uncorrected. IR spectra were measured on a
Mattson 2020 Galaxy Series FT-IR spectrometer. UV-visible
spectra were recorded in $95 \%$ ethanol on a Pharmacia Biotech
Ultraspec 2000 spectrometer. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were
recorded on a Bruker ARX250 spectrometer operating at
250.13 and 62.9 MHz , respectively; coupling constants are
reported in $\mathrm{Hz} .{ }^{13} \mathrm{C}$ Assignments ( $\mathrm{C}=$ quaternary carbon) were
based on DEPT135 and DEPT90 experiments. Mass spectra
were recorded on an AEI MS-902, a VG Micromass 7070E or a
VG Platform spectrometer. Silica gel C 60 H was used for flash
column chromatography.
9-(2-Bromophenylimino)-9,10-dihydro-10-methylacridine 3a.
9-(2-Bromoanilino)acridine ( $\mathbf{1 b}, 1.47 \mathrm{~g}, 4.22 \mathrm{mmol}$ ) (prepared
by basification of the hydrochloride salt) ${ }^{14}$ in dry THF $\left(10 \mathrm{~cm}^{3}\right)$
was added to a stirred suspension of sodium hydride ( $60 \%$ dis-
persion in mineral oil, 1.5 mol equiv.) and dry THF $\left(10 \mathrm{~cm}^{3}\right)$ at
$25^{\circ} \mathrm{C}$ under a nitrogen atmosphere. Methyl iodide $(1.20 \mathrm{~g}, 8.44$
mmol ) was added and the mixture was stirred for 24 h after
which it was quenched with water and extracted with ethyl
acetate $\left(3 \times 25 \mathrm{~cm}^{3}\right)$. Evaporation of the dried $\left(\mathrm{MgSO}_{4}\right)$
organic extract afforded the orange $N$-methylacridine 3a (1.55
$\mathrm{g}, 98 \%), \mathrm{mp} 163-164^{\circ} \mathrm{C}$; $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1597,1460,1103,772 ;$
$\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 7.97(2 \mathrm{H}, \mathrm{br} \mathrm{m}, \mathrm{H}-1,8), 7.64(1 \mathrm{H}, \mathrm{dd}, J 1.3,8.0$,
H-3'), 7.52 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3,6$ ), 7.32 ( $2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{H}-4,5$ ), 7.20
( 1 H , ddd, $J 1.4,7.4,8.8, \mathrm{H}-5^{\prime}$ ), 7.02 ( 2 H , br t, $J 7.3, \mathrm{H}-2,7$ ),
6.89 ( 1 H , ddd, $J 1.6,7.4,8.9$, H-4'), $6.75(1 \mathrm{H}, \mathrm{dd}, J 1.5,7.8$,
$\left.\mathrm{H}-6^{\prime}\right), 3.74\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 152.97$ (C), 151.46 (C),
$141.97(\mathrm{C}), 133.07(\mathrm{CH}), 131.38(\mathrm{CH}), 128.36(\mathrm{CH}), 127.87(\mathrm{C})$,
$122.85(\mathrm{CH}), 120.63(\mathrm{CH}), 119.41(\mathrm{CH}), 114.66(\mathrm{C}), 114.05$
(CH), $34.13\left(\mathrm{CH}_{3}\right)$; $m / z(\mathrm{ES}), 363.0\left(\mathrm{M}^{+}+1,100 \%\right)$ [Found: $m / z$
(HRMS-EI) $362.0436 . \mathrm{C}_{20} \mathrm{H}_{15} \mathrm{BrN}_{2}$ requires 362.0419].

9-(2-Bromophenylimino)-9,10-dihydro-10-ethylacridine 3b. Similarly prepared, from $\mathbf{1 b}$ with ethyl iodide ( $91 \%$ ), mp $131-$ $132{ }^{\circ} \mathrm{C} ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1595,1487,1467,1381,1294,1177$, 746; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.05(2 \mathrm{H}, \mathrm{br} \mathrm{m}, \mathrm{H}-1,8), 7.63(1 \mathrm{H}, \mathrm{d}, J 8.0$, H-3'), 7.54 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3,6$ ), 7.38 ( $2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{H}-4,5$ ), 7.21 ( 1 H, ddd, $\left.J 1.4,7.4,8.8, \mathrm{H}-5^{\prime}\right)$, 7.02 ( 2 H , br s, H-2,7), 6.88 ( 1 H , ddd, J 1.6, 7.4, 8.9, H-4'), 6.75 ( $1 \mathrm{H}, \mathrm{d}, ~ J 7.9, ~ H-6 '), ~ 4.33$ ( $2 \mathrm{H}, \mathrm{q}, J 8.0, \mathrm{CH}_{2}$ ), $1.55\left(3 \mathrm{H}, \mathrm{t}, J 8.0, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 152.54$ (C), 151.68 (C), 140.65 (C), $133.05(\mathrm{CH}), 128.37(\mathrm{CH}), 128.23$ (C), 122.59 (CH), $120.51(\mathrm{CH}), 119.26(\mathrm{CH}), 114.41$ (C), 113.80 $(\mathrm{CH}), 41.37\left(\mathrm{CH}_{2}\right), 12.14\left(\mathrm{CH}_{3}\right) ; m / z(\mathrm{APCI}), 377.2\left(\mathrm{M}^{+}+1\right.$, $100 \%$ ) (Found: C, 67.06; H, 4.44; N, 7.64. $\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{BrN}_{2}$ requires C, 66.86; H, 4.54; N, 7.42\%).

9-(2-Bromophenylimino)-9,10-dihydro-10-propylacridine 3c. Similarly prepared, from $\mathbf{1 b}$ with propyl iodide ( $91 \%$ ), mp 197$198{ }^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1620,1593,1487$, 1456, 1294, 1177, 746; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.05$ ( 2 H, br m, H-1,8), 7.62 $\left(1 \mathrm{H}, \mathrm{dd}, J 1.5,8.0, \mathrm{H}-3^{\prime}\right), 7.54(2 \mathrm{H}, \mathrm{t}, J 7.7, \mathrm{H}-3,6), 7.34(2 \mathrm{H}$, d, J8.7, H-4,5), 7.22 ( $\left.1 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{H}-5^{\prime}\right), 7.03(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{H}-2,7$ ), $6.89\left(1 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{H}-4^{\prime}\right), 6.80\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{H}-6^{\prime}\right), 4.17(2 \mathrm{H}, \mathrm{t}$, $J 8.4, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $1.99\left(2 \mathrm{H}\right.$, sex, $\left.J 8.0, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.16$ $\left(3 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 152.62(\mathrm{C}), 151.59(\mathrm{C})$, 140.95 (C), 133.09 (CH), 131.58 (CH), 128.39 (CH), 128.22 (C), $122.68(\mathrm{CH}), 120.57(\mathrm{CH}), 119.39(\mathrm{CH}), 114.49(\mathrm{C}), 114.05$ $(\mathrm{CH}), 48.39\left(\mathrm{CH}_{2}\right), 19.90\left(\mathrm{CH}_{2}\right), 11.11\left(\mathrm{CH}_{3}\right) ; m / z(\mathrm{APCI})$, $391.0\left(\mathrm{M}^{+}+1,100 \%\right)$ (Found: C, 67.28; H, 4.85; N, 7.34. $\mathrm{C}_{22} \mathrm{H}_{19} \mathrm{BrN}_{2}$ requires C, $67.53 ; \mathrm{H}, 4.89$; $\left.\mathrm{N}, 7.16 \%\right)$.

## Synthesis of 9-substituted acridines from 9-chloroacridines

9-(2-Bromo- N -methylanilino)acridine 7a. 2-Bromo- N -methylaniline ( $1.16 \mathrm{~g}, 6.22 \mathrm{mmol}$ ) was converted to its anion with sodium hydride in mineral oil at $25^{\circ} \mathrm{C}$ under nitrogen (see synthesis of 3a) and reacted with 9-chloroacridine ( $\mathbf{6 a}, 1.37 \mathrm{~g}, 6.22$ $\mathrm{mmol})^{15}$ in DMF at $140^{\circ} \mathrm{C}$ for 2.5 h . The cooled reaction mixture was triturated with ice-water and organic products were extracted into diethyl ether $\left(3 \times 50 \mathrm{~cm}^{3}\right)$. The dried $\left(\mathrm{MgSO}_{4}\right)$ fractions were evaporated and the residue was purified by flash column chromatography with elution by ethyl acetate-hexane ( $1: 4$ ). The yellow acridine $7 \mathrm{a}(0.66 \mathrm{~g}, 29 \%)$ had $\mathrm{mp} 153-154^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1549,1474,1402,1362$, 1099, 752; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.35(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-1,8), 8.05(2 \mathrm{H}, \mathrm{d}, J 9.0$, H-4,5), 7.77 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-2,7$ ), 7.45 ( $\left.5 \mathrm{H}, \mathrm{m}, \mathrm{H}-3,6,3^{\prime}, 4^{\prime}, 5^{\prime}\right), 6.93$ ( $\left.1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6^{\prime}\right), 3.77\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right)$; $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 151.47(\mathrm{C})$, 150.07 (C), $148.52(\mathrm{C}), 135.22(\mathrm{CH}), 130.03(\mathrm{CH}), 129.96(\mathrm{CH})$, $128.54(\mathrm{CH}), 125.63(\mathrm{CH}), 124.33(\mathrm{CH}), 124.16(\mathrm{C}), 122.88$ $(\mathrm{CH}), 119.91(\mathrm{CH}), 113.71(\mathrm{C}), 43.90\left(\mathrm{CH}_{3}\right) ; m / z(\mathrm{APCI}) 363.3$ $\left(\mathrm{M}^{+}+1,100 \%\right)$ (Found: C, $66.09 ; \mathrm{H}, 4.10 ; \mathrm{N}, 7.83 . \mathrm{C}_{20} \mathrm{H}_{15} \mathrm{BrN}_{2}$ requires C, 66.13; H, 4.16; N, 7.71\%).

1-Bromo-9-( $N$-methylanilino)acridine 7b. 1-Bromo-9-chloroacridine ( $\mathbf{6 b}, 0.48 \mathrm{~g}, 1.65 \mathrm{mmol})^{16}$ and $N$-methylaniline $(0.193 \mathrm{~g}$, $1.8 \mathrm{mmol})$ were heated in HMPA $\left(5.0 \mathrm{~cm}^{3}\right)$ at $110^{\circ} \mathrm{C}$ for 0.5 h . Addition of water liberated the acridine $7 \mathrm{~b}(0.46 \mathrm{~g}, 77 \%)$ as a yellow solid, $\mathrm{mp} 151-152^{\circ} \mathrm{C}$; $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1597,1543,1499$, 1397, 1300, 1107, 754; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.31(1 \mathrm{H}, \mathrm{dd}, J 1.2,7.6$, $\mathrm{H}-2), 8.27(1 \mathrm{H}, \mathrm{dd}, J 0.8,7.7, \mathrm{H}-5), 7.88(1 \mathrm{H}, \mathrm{dd}, J 1.2,7.3$, H-4), 7.85 ( $1 \mathrm{H}, \mathrm{dt}, J 1.0,8.9, \mathrm{H}-8), 7.79$ ( 1 H , ddd, $J 1.4,6.6$, 8.6, H-6), 7.59 ( $1 \mathrm{H}, \mathrm{dd}, J 7.3,8.7, \mathrm{H}-3$ ), 7.47 ( 1 H , ddd, $J 1.2$, 6.6, 8.7, H-7), 7.19 ( $2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{H}-3^{\prime}, 5^{\prime}$ ), 6.77 ( $1 \mathrm{H}, \mathrm{tt}, J 2.0,7.3$, $\left.\mathrm{H}-4^{\prime}\right), 6.40\left(2 \mathrm{H}, \mathrm{br} \mathrm{s},, \mathrm{H}-2^{\prime}, 6^{\prime}\right), 3.43\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right)$; $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right)$ 151.32 (C), 150.51 (C), 149.35 (C), $133.55(\mathrm{CH}), 130.92(\mathrm{CH})$, $130.79(\mathrm{CH}), 130.03(\mathrm{CH}), 129.92(\mathrm{CH}), 129.14(2 \times \mathrm{CH})$, 127.12 (CH), 126.04 (C), 124.94 (CH), 123.49 (C), $117.32(\mathrm{CH})$, $115.64(\mathrm{C}), 112.53(\mathrm{C}), 40.67\left(\mathrm{CH}_{3}\right) ; m / z(\mathrm{ES}) 363.0\left(\mathrm{M}^{+}+1\right.$, $100 \%$ ) [Found: $m / z$ (HRMS-EI) 362.0415. $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{BrN}_{2}$ requires 362.0419].

9-(Indolin-1-yl)acridine 13a. Prepared ( $60 \%$ ) from 9-chloroacridine and indoline. ${ }^{17}$

9-(1,2,3,4-Tetrahydroquinolin-1-yl)acridine 13b. Prepared ( $48 \%$ ) from 9-chloroacridine $\mathbf{6 a}(0.91 \mathrm{~g}, 4.24 \mathrm{mmol}$ ) and $1,2,3,4-$ tetrahydroquinoline ( $0.58 \mathrm{~g}, 4.33 \mathrm{mmol}$ ) in HMPA at $110^{\circ} \mathrm{C}$ (see preparation of 7b), the red acridine 13b had $\mathrm{mp} 194-195^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1555,1493,1414,1302$, 1184,$758 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.36(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{H}-4,5), 8.05(2 \mathrm{H}, \mathrm{d}$, $J 8.7, \mathrm{H}-1,8), 7.80(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3,6), 7.49(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-2,7), 7.17$ ( $\left.1 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{H}-5^{\prime}\right), 6.71\left(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-6^{\prime}, 7^{\prime}\right), 5.81(1 \mathrm{H}, \mathrm{d}, J 8.4$, H-8'), 3.82 ( $2 \mathrm{H}, \mathrm{t}, J 5.5, \mathrm{H}-2^{\prime}$ ), 3.14 ( $2 \mathrm{H}, \mathrm{t}, J 6.4, \mathrm{H}-4^{\prime}$ ), 2.34 (2H, quin, $J$ 2.9, H-3'); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 150.58$ (C), 149.33 (C), $145.09(\mathrm{C}), 130.21(\mathrm{CH}), 130.17(\mathrm{CH}), 129.29(\mathrm{CH}), 127.10$ $(\mathrm{CH}), 126.05(\mathrm{CH}), 124.74(\mathrm{C}), 124.06(\mathrm{CH}), 120.95(\mathrm{C}), 117.28$ $(\mathrm{CH}), 113.57(\mathrm{CH}), 51.38\left(\mathrm{CH}_{2}\right), 27.93\left(\mathrm{CH}_{2}\right), 22.36\left(\mathrm{CH}_{2}\right) ; \mathrm{m} / \mathrm{z}$ (ES) 311.1 ( $\mathrm{M}^{+}+1,100 \%$ ) [Found: $m / z$ (HRMS-EI) 310.1465. $\mathrm{C}_{22} \mathrm{H}_{18} \mathrm{~N}_{2}$ requires 310.1470].

9-(7-Iodoindolin-1-yl)acridine 13c. Prepared (23\%) from 9 -chloroacridine $\mathbf{6 a}$ and the sodio derivative of 7 -iodoindoline ( 1 mol equiv.) (see preparation of $7 \mathbf{7 a}$ ), the yellow acridine 13c had mp 179-181 ${ }^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1553$, 1447, 1431, 1410, 1223, 756; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.31(2 \mathrm{H}, \mathrm{d}, J 8.8$, H-4,5), 8.02 ( $2 \mathrm{H}, \mathrm{d}, J 9.0, \mathrm{H}-1,8), 7.78$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3,6$ ), 7.48 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-2,7$ ), 7.39 ( $1 \mathrm{H}, \mathrm{d}, J 8.0$, H-6'), 7.31 ( $1 \mathrm{H}, \mathrm{d}, J 6.5$, H-4'), $6.58\left(1 \mathrm{H}, \mathrm{t}, J 7.6, \mathrm{H}^{\prime} 5^{\prime}\right), 4.21\left(2 \mathrm{H}, \mathrm{t}, J 9.0, \mathrm{H}-2^{\prime}\right), 3.47$ $\left(2 \mathrm{H}, \mathrm{t}, J 9.0, \mathrm{H}-3^{\prime}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 151.69(\mathrm{C}), 149.89(\mathrm{C}), 147.84$ (C), 139.57 (CH), 132.24 (C), $130.80(\mathrm{CH}), 130.25(\mathrm{CH}), 126.35$ $(\mathrm{CH}), 125.82(\mathrm{C}), 125.56(\mathrm{CH}), 124.46(\mathrm{CH}), 121.90(\mathrm{CH})$, $74.53(\mathrm{C}), 57.50\left(\mathrm{CH}_{2}\right), 30.06\left(\mathrm{CH}_{2}\right) ; m / z(\mathrm{APCI}) 423.5\left(\mathrm{M}^{+}+\right.$ 1, $100 \%$ ) (Found: C, $59.72 ; \mathrm{H}, 3.47$; N, 6.51. $\mathrm{C}_{21} \mathrm{H}_{15} \mathrm{IN}_{2}$ requires C, 59.73; H, 3.58; N, 6.63\%).

1-Bromo-9-(indolin-1-yl)acridine 13d. Prepared (83\%) from 1-bromo-9-chloroacridine $\mathbf{6 b}$ and indoline in HMPA (see preparation of 7b), the acridine $\mathbf{1 3 d}$ had $\mathrm{mp} 168-169^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1601,1493,1425,1406,1263,762$, $735 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.30(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{H}-2,5), 8.00(1 \mathrm{H}, \mathrm{d}, J 8.7$, H-8), 7.88 ( $1 \mathrm{H}, \mathrm{d}, J 7.3, \mathrm{H}-4$ ), $7.80(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 7.58(1 \mathrm{H}, \mathrm{dd}$, $J 7.3,8.7, \mathrm{H}-3), 7.46(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7), 7.26\left(1 \mathrm{H}, \mathrm{d}, J 7.2, \mathrm{H}-4^{\prime}\right)$, 6.92 (1 H, t, J 7.2, H-6'), 6.52 ( $\left.1 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{H}-5^{\prime}\right), 5.84(1 \mathrm{H}, \mathrm{d}$, $\left.J 7.7, \mathrm{H}^{\prime} 7^{\prime}\right), 4.20\left(2 \mathrm{H}, \mathrm{d}, J 9.4, \mathrm{H}^{\prime} 2^{\prime}\right), 3.50(2 \mathrm{H}$, br m, H-3'); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 151.91(\mathrm{C}), 151.15(\mathrm{C}), 150.25(\mathrm{C}), 145.65(\mathrm{C})$, $133.39(\mathrm{CH}), 130.85(\mathrm{CH}), 130.69(\mathrm{CH}), 130.37(\mathrm{CH}), 129.75$ $(\mathrm{CH}), 128.68(\mathrm{C}), 127.62(\mathrm{CH}), 126.57(\mathrm{CH}), 125.67(\mathrm{C}), 124.80$ $(\mathrm{CH}), 124.65(\mathrm{CH}), 124.34(\mathrm{C}), 117.38(\mathrm{CH}), 116.28(\mathrm{C}), 106.65$ $(\mathrm{CH}), 55.73\left(\mathrm{CH}_{2}\right), 28.80\left(\mathrm{CH}_{2}\right) ; m / z(\mathrm{APCI}) 375.3\left(\mathrm{M}^{+}+1\right.$, $100 \%$ ) [Found: $m / z$ (HRMS-EI) 374.0409. $\mathrm{C}_{21} \mathrm{H}_{15} \mathrm{BrN}_{2}$ requires 374.0419].

1-Bromo-9-(1,2,3,4-tetrahydroquinolin-1-yl)acridine 13e. Prepared ( $43 \%$ ) from 1-bromo-9-chloroacridine $\mathbf{6 b}$ and 1,2,3,4tetrahydroquinoline in HMPA (see preparation of 7b), this red bromoacridine 13 e had $\mathrm{mp} 164-165{ }^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1601,1543,1491,1420,1398,1302,765$; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.29(1 \mathrm{H}, \mathrm{dd}, J 1.3,8.7, \mathrm{H}-2), 8.27(1 \mathrm{H}, \mathrm{dd}, J 0.5$, 8.8, H-5), 7.97 ( 1 H , ddd, $J 0.8,1.5,8.9, \mathrm{H}-8$ ), 7.89 ( 1 H , dd, $J 1.2,7.2, \mathrm{H}-4), 7.79$ ( 1 H , ddd, $J 1.4,6.6,8.8, \mathrm{H}-6$ ), 7.58 ( 1 H , dd, $J 7.2,8.7, \mathrm{H}-3), 7.47$ ( 1 H , ddd, $J 1.1,6.5,8.7, \mathrm{H}-7$ ), 7.14 ( $\left.1 \mathrm{H}, \mathrm{d}, J 7.3, \mathrm{H}-5^{\prime}\right), 6.79(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7$ '), $6.65(1 \mathrm{H}, \mathrm{dt}, J 1.2$, 6.0, H-6'), 5.74 ( $1 \mathrm{H}, \mathrm{d}, J$ 8.2, H-8'), 3.72 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-2^{\prime}$ ), 3.07 ( $\left.2 \mathrm{H}, \mathrm{t}, J 6.2, \mathrm{H}-4^{\prime}\right), 2.31\left(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3^{\prime}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 151.37(\mathrm{C})$, 150.48 (C), 148.96 (C), 145.95 (C), $133.52(\mathrm{CH}), 131.02(\mathrm{CH})$, $130.62(\mathrm{CH}), 129.97(\mathrm{CH}), 129.17(\mathrm{CH}), 128.98(\mathrm{CH}), 127.27$ $(\mathrm{CH}), 127.11(\mathrm{CH}), 125.55(\mathrm{C}), 124.77(\mathrm{CH}), 124.13(\mathrm{C}), 117.04$ $(\mathrm{CH}), 116.10(\mathrm{C}), 114.06(\mathrm{CH}), 52.43\left(\mathrm{CH}_{2}\right), 27.89\left(\mathrm{CH}_{2}\right), 21.60$ $\left(\mathrm{CH}_{2}\right) ; m / z(\mathrm{ES}) 388.9\left(\mathrm{M}^{+}+1,100 \%\right)$ [Found: $m / z$ (HRMS-EI) 388.0573. $\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{BrN}_{2}$ requires 388.0575].

1-Bromo-9-(6-nitroindolin-1-yl)acridine 13f. Formed (79\%) from 1-bromo-9-chloroacridine 6b and 6-nitroindoline ( 1 mol equiv.) in HMPA (see preparation of 7b), the yellow
nitroindolinylacridine 13f had mp $245-246{ }^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1493,1427,1337,1262,1098,760$; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.33(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{H}-2,8), 7.93(1 \mathrm{H}, \mathrm{dd}, J 1.2,7.3$, H-4), 7.86 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-5,7$ ), 7.63 ( $1 \mathrm{H}, \mathrm{dd}, J 7.3,8.7, \mathrm{H}-3$ ), 7.60 ( 1 H , dd, $J 2.2,8.1, \mathrm{H}-5 '$ ), 7.54 ( 1 H , ddd, $J 1.2,6.6,8.8, \mathrm{H}-6$ ), $7.32\left(1 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{H}-4^{\prime}\right), 6.48\left(1 \mathrm{H}, \mathrm{d}, J 2.1, \mathrm{H}-7^{\prime}\right), 4.30(2 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{H}-2^{\prime}\right), 3.57\left(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3^{\prime}\right)$; $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 153.04(\mathrm{C}), 151.14$ (C), 150.28 (C), 148.92 (C), 136.47 (C), 133.98 (CH), 131.08 $(2 \times \mathrm{CH}), 130.72(\mathrm{CH}), 129.95(\mathrm{CH}), 127.56(\mathrm{CH}), 125.10(\mathrm{C})$, $124.62(\mathrm{CH}), 123.78(\mathrm{C}), 123.64(\mathrm{CH}), 115.48(\mathrm{C}), 113.48(\mathrm{CH})$, $100.19(\mathrm{CH}), 56.04\left(\mathrm{CH}_{2}\right), 28.58\left(\mathrm{CH}_{2}\right) ; m / z(\mathrm{APCI}) 420.2\left(\mathrm{M}^{+}\right.$ $+1,100 \%$ ) (Found: C, 60.33; H, 3.34; N, 9.95. $\mathrm{C}_{21} \mathrm{H}_{14} \mathrm{BrN}_{3} \mathrm{O}_{2}$ requires $\mathrm{C}, 60.02 ; \mathrm{H}, 3.36 ; \mathrm{N}, 10.00 \%$ ).

1-Bromo-9-(2-methylindolin-1-yl)acridine (16: mixture of stereoisomers). Prepared ( $78 \%$ ) from 1-bromo-9-chloroacridine $\mathbf{6} \mathbf{b}$ and $( \pm)-2$-methylindoline in HMPA (see preparation of 7b), the acridine 16 was isolated as a mixture of diastereomers; $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1603,1543,1485,1422,1404,1262,741$; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.69(6 \mathrm{H}, \mathrm{m}), 8.12(2 \mathrm{H}, \mathrm{d}, J 8.9), 7.89(2 \mathrm{H}, \mathrm{d}$, $J 7.2$ ), 7.85 ( $2 \mathrm{H}, \mathrm{d}, J 7.2$ ), $7.79(3 \mathrm{H}, \mathrm{m}), 7.57(3 \mathrm{H}, \mathrm{m}), 7.45$ $(3 \mathrm{H}, \mathrm{m}), 7.26(3 \mathrm{H}, \mathrm{m}), 6.97(3 \mathrm{H}, \mathrm{q}, J 7.7), 6.43(3 \mathrm{H}, \mathrm{m}), 6.04$ ( $2 \mathrm{H}, \mathrm{d}, J 7.8$ ), $5.88(1 \mathrm{H}, \mathrm{d}, J 7.8), 4.69(3 \mathrm{H}, \mathrm{m}), 3.79(2 \mathrm{H}, \mathrm{dd}$, $J 10.3,16.1), 3.45(1 \mathrm{H}, \mathrm{dd}, J 8.7,15.6), 3.16(1 \mathrm{H}, \mathrm{dd}, J 11.6$, 15.7 ), 2.98 ( $2 \mathrm{H}, \mathrm{dd}, J 4.7,16.1$ ), 1.21 ( $3 \mathrm{H}, \mathrm{d}, J 6.2$ ), 0.91 ( 6 H , d, $J 6.4$ ); $m / z(\mathrm{ES}) 388.6\left(\mathrm{M}^{+}+1,100 \%\right.$ ) [Found: $m / z$ (HRMSEI) 388.0586. $\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{BrN}_{2}$ requires 388.0575].

## General method for the radical synthesis of penta- and hexacyclic acridines

A solution of tributyltin hydride ( 1 mol equiv.) and AIBN ( 0.1 mol equiv.) was added to a boiling solution of the substituted anilinoacridine in dry toluene under nitrogen over 1 h . The mixture was refluxed ( 12 h ) and the product isolated by vacuum evaporation of solvent followed by column chromatography.
$\mathbf{8 H}$-Quino[4,3,2-kl]acridine 2. Cyclisation of 9-(2-bromoanilino) acridine $\mathbf{1 b}^{14}$ with tributyltin hydride-AIBN, followed by chromatographic fractionation with ethyl acetate-hexane ( $1: 2$ ) gave recovered starting material ( $68 \%$ ) and the quinoacridine ( $31 \%$ ), identical (UV, IR, ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR) to an authentic sample. ${ }^{5}$

8-Methyl-8H-quino[4,3,2-kl]acridine 4a. Formed from 3a, following chromatographic fractionation with ethyl acetatehexane ( $1: 4$ to $1: 1$ gradient), the methylquinoacridine $\mathbf{4 a}$ ( $69 \%$ ) had mp 210-211 ${ }^{\circ} \mathrm{C}$ (from ethyl acetate) (lit. ${ }^{8} \mathrm{mp} 208-$ $210^{\circ} \mathrm{C}$, lit. ${ }^{9} \mathrm{mp} 204-206^{\circ} \mathrm{C}$ ); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1589,1557,1491$, $\left.1458,1359,745 ; \delta_{\mathrm{H}}\left({ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 8.78(1 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{H}-12)$, $8.54(1 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{H}-4), 8.14(1 \mathrm{H}, \mathrm{d}, J 7.9, \mathrm{H}-5), 7.91(1 \mathrm{H}$, dd, $J 1.2,8.2, \mathrm{H}-1), 7.86(1 \mathrm{H}, \mathrm{t}, J 8.2, \mathrm{H}-6), 7.63(3 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-2,9,10), 7.50(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 7.38$ ( $1 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{H}-7$ ), 7.25 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-11$ ), $\left.3.70\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left({ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 149.30(\mathrm{C})$, 145.22 (C), 141.47 (C), 141.40 (C), 134.06 (C), 132.48 (CH), $132.29(\mathrm{CH}), 129.45(\mathrm{CH}), 128.82(\mathrm{CH}), 125.34(\mathrm{CH}), 125.21$ $(\mathrm{CH}), 123.29(\mathrm{CH}), 122.90(\mathrm{C}), 121.51(\mathrm{CH}), 121.39(\mathrm{C}), 115.89$ (C), $115.15(\mathrm{CH}), 111.57(\mathrm{CH}), 33.61\left(\mathrm{CH}_{3}\right) ; m / z(\mathrm{APCI})$ $283.5\left(\mathrm{M}^{+}+1,100 \%\right)$ [Found: $m / z$ (HRMS-FAB) 283.1212. $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{~N}_{2}$ requires 283.1235].

8-Ethyl-8H-quino[4,3,2-kl] acridine 4b. Similarly prepared from 3b, the ethylquinoacridine $\mathbf{4 b}(79 \%)$ had $\mathrm{mp} 176-177^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1595,1491,1460,1375$, 1101,$737 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.97(1 \mathrm{H}, \mathrm{d}, J 7.7, \mathrm{H}-12), 8.38(1 \mathrm{H}, \mathrm{dd}$, $J 1.3,8.0, \mathrm{H}-4), 8.04(1 \mathrm{H}, \mathrm{dd}, J 1.3,8.0, \mathrm{H}-1), 7.98(1 \mathrm{H}, \mathrm{d}$, $J 8.0, \mathrm{H}-5), 7.76(1 \mathrm{H}, \mathrm{t}, J 8.2, \mathrm{H}-6), 7.66(1 \mathrm{H}$, ddd, $J 1.3,7.0$, $8.3, \mathrm{H}-2), 7.56(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-10)$, $7.48(1 \mathrm{H}$, ddd, $J 1.3,7.0,8.2$, H-3), 7.28 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-9,11$ ), 7.11 ( $1 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{H}-7$ ), $4.24(2 \mathrm{H}$, $\left.\mathrm{q}, J 7.2, \mathrm{CH}_{2}\right), 1.52\left(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 149.76(\mathrm{C})$,
145.48 (C), 140.60 (C), 140.44 (C), 134.90 (C), $131.72(\mathrm{CH})$, $129.06(\mathrm{CH}), 126.30(\mathrm{CH}), 124.81(\mathrm{CH}), 123.11(\mathrm{C}), 122.53$ $(2 \times \mathrm{CH}), 122.10(\mathrm{C}), 121.43(\mathrm{CH}), 116.45(\mathrm{C}), 113.35$ $(2 \times \mathrm{CH}), 111.20(\mathrm{CH}), 107.54(\mathrm{CH}), 41.01\left(\mathrm{CH}_{2}\right), 11.35\left(\mathrm{CH}_{3}\right)$; $\mathrm{m} / \mathrm{z}$ (APCI) $297.3\left(\mathrm{M}^{+}+1,100 \%\right)$ [Found: $m / z$ (HRMS-FAB) 297.1384. $\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{~N}_{2}$ requires 297.1392].

8-Propyl-8H-quino[4,3,2-kl]acridine 4c. Similarly prepared from 3c, the propylquinoacridine $4 \mathbf{c}(54 \%)$ had $\mathrm{mp} 194-196^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1591,1555,1491,1458$, 1371, 745 ; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.97(1 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{H}-12), 8.38(1 \mathrm{H}, \mathrm{dd}$, $J 1.3,8.1, \mathrm{H}-4), 8.04(1 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{H}-1), 7.98(1 \mathrm{H}, \mathrm{d}, J 7.9$, H-5), $7.76(1 \mathrm{H}, \mathrm{t}, J 8.2, \mathrm{H}-6), 7.66(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2), 7.55(1 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-10)$, $7.47(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 7.26(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-9,11), 7.08(1 \mathrm{H}, \mathrm{d}$, $J 8.2, \mathrm{H}-7), 4.08\left(2 \mathrm{H}, \mathrm{t}, J 8.4, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.96(2 \mathrm{H}$, sex, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $1.18\left(3 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 149.71$ (C), 145.56 (C), 140.77 (C), 140.62 (C), 134.76 (C), 131.60 $(\mathrm{CH}), 129.00(\mathrm{CH}), 126.17(\mathrm{CH}), 124.74(\mathrm{CH}), 123.06(\mathrm{C})$, $122.50(2 \times \mathrm{CH}), 122.10(\mathrm{C}), 121.34(\mathrm{CH}), 116.41(\mathrm{C}), 113.51$ $(2 \times \mathrm{CH}), 111.11(\mathrm{CH}), 107.72(\mathrm{CH}), 47.88\left(\mathrm{CH}_{2}\right), 19.09\left(\mathrm{CH}_{2}\right)$, $11.12\left(\mathrm{CH}_{3}\right) ; m / z(\mathrm{APCI}) 311.2\left(\mathrm{M}^{+}+1,100 \%\right)$ [Found: $m / z$ (HRMS-FAB) 311.1557. $\mathrm{C}_{22} \mathrm{H}_{19} \mathrm{~N}_{2}$ requires 311.1584].

13-Methyl-13H-quino[4,3,2-kl] acridine 10. Formed from 7a, following chromatographic fractionation with ethyl acetate, the 13-methylquinoacridine ( $50 \%$ ) had mp $118-119{ }^{\circ} \mathrm{C} ; v_{\text {max }}(\mathrm{KBr}) /$ $\mathrm{cm}^{-1} 1611,1562,1541,1466,1362,1150,754 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.07$ $(4 \mathrm{H}, \mathrm{m}, \mathrm{H}-1,4,5,12), 7.80(1 \mathrm{H}, \mathrm{t}, J 7.6, \mathrm{H}-6), 7.75(1 \mathrm{H}, \mathrm{d}, J 6.9$, H-7), 7.65 ( $1 \mathrm{H}, \mathrm{ddd}, J 1.4,6.7,8.7, \mathrm{H}-2), 7.50(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-9,10)$, 7.34 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-11$ ), 7.29 ( 1 H , ddd, J 1.4, 6.7, 8.8, H-3), 4.09 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}$ ); $\delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 152.26(\mathrm{C}), 149.44(\mathrm{C}), 147.85$ (C), $142.80(\mathrm{C}), 131.65(\mathrm{CH}), 130.91(\mathrm{CH}), 130.74(\mathrm{CH}), 130.66$ (C), $129.64(\mathrm{CH}), 127.16(\mathrm{CH}), 125.14(\mathrm{CH}), 124.70(\mathrm{CH})$, $124.36(\mathrm{C}), 124.14(\mathrm{CH}), 121.95(\mathrm{CH}), 118.90(\mathrm{CH}), 117.94(\mathrm{C})$, $115.06(\mathrm{C}), 111.93(\mathrm{CH}), 45.74\left(\mathrm{CH}_{3}\right) ; m / z(\mathrm{ES}) 283.5\left(\mathrm{M}^{+}+1\right.$, $100 \%$ ) [Found: $m / z$ (HRMS-FAB) 283.1239. $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{~N}_{2}$ requires 283.1235].

The same 13-methylquinoacridine ( $56 \%$ ) was formed from 7b and tributyltin hydride-AIBN in boiling toluene.

## 13,14-Dihydropyrrolo $\left.{ }^{\prime} 3^{\prime}, 2^{\prime}, 1^{\prime}: 8,1\right]$ quino[4,3,2-kl] acridine

14a. Formed from 13c ( $62 \%$ ), as a red solid (from ethyl acetate), $\mathrm{mp}>100{ }^{\circ} \mathrm{C}$ (decomp.); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1}$ 2926, 1561, 1526, 1458, 1418, 1343, 772; $\left.\delta_{\mathrm{H}}\left({ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 8.83$ ( $1 \mathrm{H}, \mathrm{d}, J 9.4$, H-11), 8.31 ( $1 \mathrm{H}, \mathrm{d}, J 6.7, \mathrm{H}-3$ ), 8.19 ( $1 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{H}-4$ ), 8.07 $(1 \mathrm{H}, \mathrm{t}, J 8.0, \mathrm{H}-5), 7.91(1 \mathrm{H}, \mathrm{t}, J 7.6, \mathrm{H}-9), 7.68(2 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-1,8), 7.60(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-2,6), 7.46(1 \mathrm{H}, \mathrm{t}, J 8.2, \mathrm{H}-10), 5.35$ ( $2 \mathrm{H}, \mathrm{t}, J 6.9, \mathrm{H}-13$ ). $3.64(2 \mathrm{H}, \mathrm{t}, J 6.9, \mathrm{H}-14)$; $\delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right)$ 147.77 (C), 142.62 (C), 141.09 (C), 140.03 (C), $135.50(\mathrm{CH})$, $135.42(\mathrm{CH}), 134.12(\mathrm{C}), 131.89(\mathrm{C}), 129.18(\mathrm{CH}), 128.39(\mathrm{CH})$, $127.01(\mathrm{CH}), 123.80(\mathrm{CH}), 121.67(\mathrm{C}), 121.07(\mathrm{CH}), 119.82$ (CH), 115.68 (C), 115.07 (CH), 113.48 (CH), 113.17 (C), 57.47 $\left(\mathrm{CH}_{2}\right), 28.73\left(\mathrm{CH}_{2}\right) ; m / z(\mathrm{APCI}) 295.4\left(\mathrm{M}^{+}+1,100 \%\right)$ [Found: $m / z$ (HRMS-FAB) 295.1234. $\mathrm{C}_{21} \mathrm{H}_{15} \mathrm{~N}_{2}$ requires 295.1235].

The same hexacyclic acridine $\mathbf{1 4 a}(22 \%)$ was formed from 13d and tributyltin hydride-AIBN in boiling toluene.

14,15-Dihydro-13H-pyrido[ $3^{\prime}, 2^{\prime}, 1^{\prime}: 8,1$ ]quino[4,3,2-kl]acridine 14b. Formed ( $65 \%$ ) from 13e after purification by flash column chromatography (ethyl acetate-hexane, methanol), this hexacycle had mp 215-216 ${ }^{\circ} \mathrm{C}$; $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 2365,1557,1541$, $1429,1371,1138,752 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.14(1 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{H}-11)$, $8.04(1 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{H}-8), 7.96(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3,4), 7.79(1 \mathrm{H}, \mathrm{t}, J 7.9$, H-5), 7.73 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ ), 7.63 ( $1 \mathrm{H}, \mathrm{ddd}, J 1.3,6.7,8.2$, H-9), 7.30 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-1,2$ ), 7.25 ( 1 H, ddd, $J$ 1.2, 6.6, $8.0, \mathrm{H}-10$ ), 3.59 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-13$ ), $3.21(2 \mathrm{H}, \mathrm{t}, J 6.5, \mathrm{H}-15), 2.42(2 \mathrm{H}$, quin, $J 6.0$ $\mathrm{H}-14)$; $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 147.37(\mathrm{C}), 140.37$ (C), 137.72 (C), 131.58 $(\mathrm{CH}), 130.69(\mathrm{CH}), 130.13(\mathrm{CH}), 129.95(\mathrm{C}), 127.32(\mathrm{C}), 125.08$ $(\mathrm{CH}), 124.12(\mathrm{CH}), 123.85(\mathrm{C}), 121.35(\mathrm{CH}), 121.25(\mathrm{CH})$, $117.10(\mathrm{C}), 113.60(\mathrm{C}), 111.44(\mathrm{CH}), 53.12\left(\mathrm{CH}_{2}\right), 27.46\left(\mathrm{CH}_{2}\right)$,
$22.88\left(\mathrm{CH}_{2}\right) ; m / z(\mathrm{ES}) 309.2\left(\mathrm{M}^{+}+1,100 \%\right)$ [Found: $m / z$ (HRMS-FAB) 309.1387. $\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{~N}_{2}$ requires 309.1392].
( $\pm$ )-12-Methyl-12,13-dihydropyrrolo $\left[3^{\prime}, 2^{\prime}, 1^{\prime}: 8,1\right]$ quino[4,3,2$k l$ ]acridine 17. Formed from $16(28 \%) ; v_{\text {max }}\left(\mathrm{KBr}^{2} / \mathrm{cm}^{-1} 1555\right.$, $1526,1458,1410,1370,1140,760 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 8.23(1 \mathrm{H}, \mathrm{d}$, $J 8.8, \mathrm{H}-11), 7.90(1 \mathrm{H}, \mathrm{dd}, J 1.1,8.7), 7.73(2 \mathrm{H}, \mathrm{m}), 7.64(1 \mathrm{H}$, dd, $J 7.2,8.5), 7.55(1 \mathrm{H}$, ddd, $J 0.9,6.5,8.8), 7.49(1 \mathrm{H}$, dd, $J 1.2,7.2), 7.15(2 \mathrm{H}, \mathrm{m}), 5.68$ ( 1 H , quin, $J 6.6, \mathrm{H}-13$ ), 3.64 ( $1 \mathrm{H}, \mathrm{dd}, J 8.7,16.4, \mathrm{H}-14), 2.97(1 \mathrm{H}, \mathrm{d}, J 16.1, \mathrm{H}-14), 1.36$ $\left(3 \mathrm{H}, \mathrm{d}, J 6.3, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 151.16(\mathrm{C}), 150.52(\mathrm{C}), 143.67$ (C), 140.78 (C), $130.56(\mathrm{CH}), 130.50(\mathrm{CH}), 129.78(\mathrm{CH}), 129.59$ $(\mathrm{CH}), 128.88(\mathrm{C}), 124.74(\mathrm{CH}), 124.57(\mathrm{CH}), 124.24(\mathrm{CH})$, $121.56(\mathrm{C}), 121.48(\mathrm{CH}), 120.04(\mathrm{CH}), 118.28(\mathrm{C}), 112.43(\mathrm{C})$, $109.93(\mathrm{CH}), 60.69(\mathrm{CH}), 36.46\left(\mathrm{CH}_{2}\right), 20.46\left(\mathrm{CH}_{3}\right) ; \mathrm{m} / \mathrm{z}(\mathrm{APCI})$ $309.1\left(\mathrm{M}^{+}+1,100 \%\right)$ [Found: $m / z$ (HRMS-FAB) 309.1385. $\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{~N}_{2}$ requires 309.1392].

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